

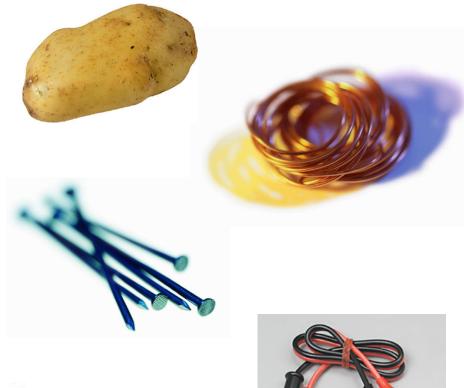


Potato Battery Activity Part 1: Engage

- How does a battery power something?
- What is a battery made of?
- What happens inside the battery that allows the electricity to come out of it?
- Why does a battery stop working? What gets used up?
- What kind of energy does a battery contain and what kind does it create?
 - Chemical, electrical, mechanical?

Part 2: Explore

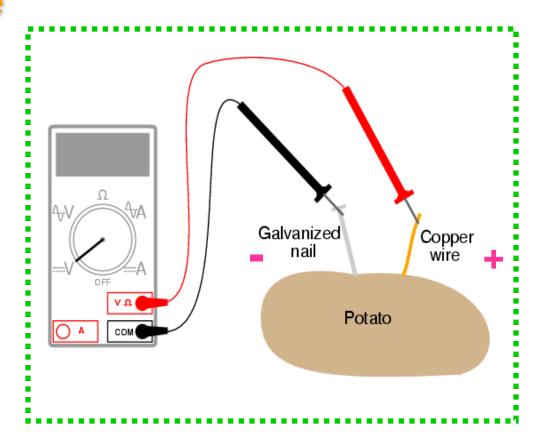
- Supplies:
 - 2 potatoes
 - 2 pieces of copper wire
 - 2 nails
 - 3 alligator clips
 - 1 LED





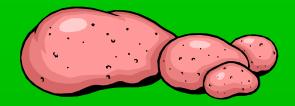


- Part 2: Explore
- Set-up:
 - Connect copper wire to + end and nail to – end of LED
 - Does the LED light up?



→How can we get more voltage?

Part 2: Explore

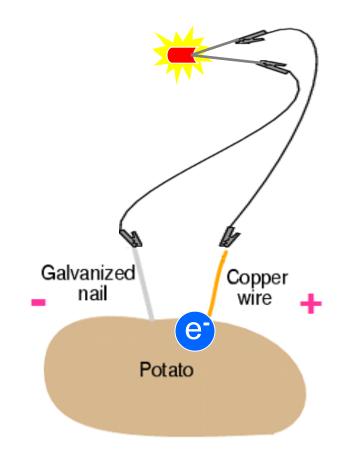


- Now make a series of connected batteries
 - How many potatoes do you need to power the LED?
- What happens when you:
 - Change the direction of the LED?
 - Use only nails or only pennies?
 - Connect the potatoes in parallel?

Potato Battery Explanation

Part 3: Explain

- Does the potato actually power the clock?
- What happens to the copper wire and zinc nail over time?



Potato Battery Explanation

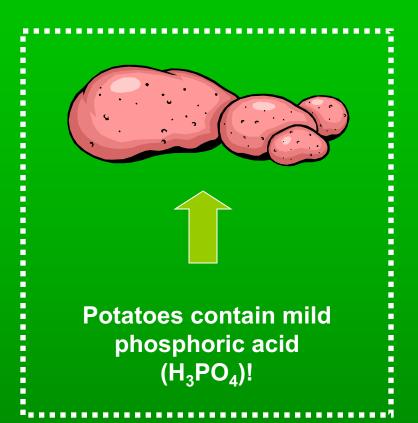
Part 3: Explain

- What is the chemistry behind the reaction?
 - Oxidation at the zinc electrode:

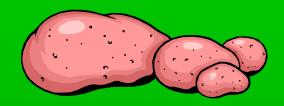
 $Zn(s) \rightarrow Zn^{2+}(aq) + 2e^{-}$

Reduction at the copper electrode:

 $Cu^{2+}(aq) + 2e^{-} \rightarrow Cu(s)$ $2H^{+}(aq) + 2e^{-} \rightarrow H_{2}(g)$

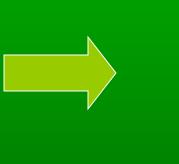


Part 4: Evaluate



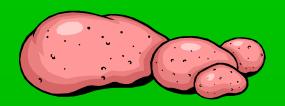
- Could this work with other vegetables or fruits?
- Is a potato a good alternate energy source?







Part 5: Elaborate



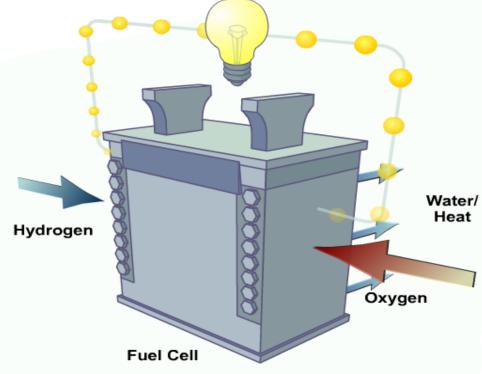
 What makes a battery stop working? What gets used up?





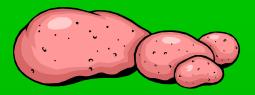
Part 5: Elaborate

- What else uses this type of chemical energy?
 - All batteries
 - Fuel cells



http://www2.wwnorton.com/colle ge/chemistry/gilbert/tutorials/ch 17.htm

Part 6: Summarize



Describe the energy transformation

